Effectiveness of passive ultrasonic irrigation on calcium hydroxide removal with different solutions

Efetividade da irrigação ultrassônica passiva na remoção de hidróxido de cálcio com diferentes soluções

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Objective: this study aimed to assess the effectiveness of Passive Ultrasonic Irrigation (PUI) with different solutions for calcium hydroxide (Ca(OH)₂) removal. Materials and method: forty-eight single-rooted mandibular premolars were prepared and filled with Ca(OH)₂. After 15 days, the specimens were randomly divided into four groups according to the method of removal of the intracanal medication (n=12). In Group 1, Ca(OH)₂ removal was performed through conventional irrigation with 2.5% NaOCl and 17% EDTA combined with the use of the 'master apical file' at working length. Groups 2 and 3 were treated in the same manner as Group 1, following the PUI with 2.5% NaOCl and 17% EDTA, and 2.5% NaOCl and 10% citric acid, respectively. Group 4 served as a control group and received no intracanal medication. The roots were longitudinally grooved and split in two halves. The half with the higher remaining amount was analyzed with a stereomicroscope, and the percentage of surface area covered by the remaining Ca(OH)₂ in relation to the total surface area and the surface area of each third of the root canal was calculated. The results were statistically analyzed with the Kruskal-Wallis Test and Mann-Whitney Test. Null hypothesis was set as 5%. Results: none of the groups tested showed similarity to the control group (P < 0.05). However, no statistical difference was observed between the removal techniques of Ca(OH)₂ from root canals. Conclusions: neither syringe irrigation nor the combination of syringe irrigation with PUI were effective methods for removing Ca(OH)₂ from the root canals.

Keywords: Calcium hydroxide. Ultrasound. Citric acid. Irrigation. Root canal.

Introduction

The reduction or elimination of bacteria and their byproducts from the root canal system is one of the main objectives of root canal treatment¹. Intracanal medication with strong antibacterial activity against most of the bacterial strains identified in root canal infections is used in order to improve disinfection of the root canal². The most commonly used intracanal medication is calcium hydroxide (Ca(OH)₂) due to its effectiveness against the majority of endodontic pathogens and its biocompatibility³.
However, Ca(OH)$_2$-based medication should be removed before root filling because Ca(OH)$_2$ residue on the canal walls negatively affects the quality of the root filling. Furthermore, remnants of Ca(OH)$_2$ influence dentine bond strength, interfere with the sealing ability, and prevent the penetration of sealers into dentinal tubules, resulting in a potential reduction of sealer adaptation.

The removal of Ca(OH)$_2$ residue from irregular canal walls is difficult. Various products and techniques were described. The most common method for removing Ca(OH)$_2$ is the use of a ‘master apical file’ (MAF) at working length combined with copious sodium hypochlorite (NaOCl) and ethylenediaminetetraacetic acid (EDTA) irrigation. With the addition of EDTA, the citric acid has also been tested. According to Rödig et al. (2010), citric acid and EDTA showed better results than NaOCl and water. Moreover, the combination of chelating agents and NaOCl did not lead to significant improvement in calcium hydroxide removal.

Nevertheless, canal irregularities may be inaccessible for conventional irrigation procedures, and Ca(OH)$_2$ may remain in these extensions. The passive ultrasonic irrigation (PUI) is effective in dentine debris removal from the root canal walls and in removing Ca(OH)$_2$ paste from the root canal walls. Saline solution, water, NaOCl, EDTA, and combinations thereof, have been used as irrigation material during PUI. To this date, no study in literature assesses the use of PUI with citric acid for Ca(OH)$_2$ removal.

Thus, this study aims to assess the effectiveness of PUI with NaOCl and different chelating solution after the use of ‘master apical file’ for Ca(OH)$_2$ removal in oval canals.

Materials and method

This study was submitted and approved by the Ethical Committee of the Federal University of Santa Maria (CAAE: 08509712.7.0000.5346).

Tooth selection

Forty-eight single-rooted mandibular premolar teeth with complete root formation, straight root, and oval cross-section of the roots were selected and stored in distilled water until use. Radiographs in the mouth-lingual and mesial-distal directions were taken and the canal was identified as oval when the mouth-lingual diameter was ≥ 2 of the mesiodistal diameter. The curvature angle of the root canal was measured following Schneider’s methodology.

Cleaning and shaping

Coronal access was performed using 1014 diamond burs and Endo-Z burs (KG Sorensen, Cotia, SP, Brazil) in water-cooled, high-speed handpiece. The clinical crown was not removed in order to allow a greater amount of irrigating solution in contact with the channel walls. The canal patency was established with a size 10 K-file (Dentsply Maillefer, Ballaigues, Switzerland). A size 15 K-file (Dentsply Maillefer) was placed into the canal until visible at the apical foramen. The working length (WL) was determined by subtracting 1 mm from this measurement.

Root canals were prepared with ProTaper rotating instruments (Dentsply Maillefer) by using a torque-control X-Smart motor (Dentsply Maillefer) rotating at 300 rpm. Initially, the cervical and middle thirds of the roots were prepared using S1, SX and S2 instruments. Later, S1, S2, F1, F2, and F3 files were sequentially used for the entire WL. Each canal was irrigated with 3 mL of a freshly prepared 2.5% NaOCl solution between each instrument change. To ensure that the apical diameter was uniform after preparation, a size 30 K-file, 0.02 taper (Dentsply Maillefer) was inserted through the entire WL. After conclusion of instrumentation, the PUI was performed with 2.5% NaOCl and 17% EDTA with a size 15 K-file, 0.02 taper (Dentsply Maillefer) in order to effectively clean the canals and remove remaining debris. The roots were subsequently rinsed with 2 mL of distilled water. The canals were aspirated with Capillary Tips (Ultradent; Munich, Germany), and dried using size 30 paper points (Dentsply Maillefer).

The Ca(OH)$_2$ paste (Calen, SS White, Rio de Janeiro, Brazil) was placed into the entire WL with a screwable syringe (Duflex, SS White, Rio de Janeiro, Brazil). A radiograph was taken to confirm the complete filling of the canal. The access cavities were temporarily sealed with a cotton pellet and Cavit (Espe, Seefeld, Germany) down to a depth of 2 mm. The roots were then placed in a sponge saturated with natural water and incubated in 100% relative humidity at 37°C for 15 days. After this period, the temporary filling material was removed using 1012 diamond burs (KG Sorensen) in water-cooled, high-speed handpiece. The medication was removed from all canals initially with 8 ml of 2.5% NaOCl, alternating with a size 30 K-file (Dentsply Maillefer) inserted in the WL with an oscillatory motion. The specimens were randomly assigned by a Random Allocation Software program (Microsoft, Washington, United States) to 3 experimental groups (n=12), according to the Ca(OH)$_2$ removal technique. The control group (n=12) received no intracanal medication (Table 1).
Table 1 – Experimental groups according to irrigation material and irrigation technique.

<table>
<thead>
<tr>
<th>Group</th>
<th>n</th>
<th>Irrigant</th>
<th>Irrigation Technique</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>12</td>
<td>NaOCl + EDTA</td>
<td>Conventional</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td>Irrigation</td>
</tr>
<tr>
<td>2</td>
<td>12</td>
<td>NaOCl + EDTA</td>
<td>PUI</td>
</tr>
<tr>
<td>3</td>
<td>12</td>
<td>NaOCl + Citric Acid</td>
<td>PUI</td>
</tr>
<tr>
<td>4</td>
<td>12</td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

Fonte: dos autores.

In Group 1, final irrigation was performed with 2 mL of 2.5% NaOCl and 2 mL of 17% EDTA, manually agitated with a size 15 K-file, 0.02 taper (Dentsply Maillefer) for 3 minutes with an oscillatory motion. Finally, root canals were irrigated with 2 mL of 2.5% NaOCl to neutralize the action of the EDTA.

In Group 2, root canals were irrigated with PUI using the piezoelectric unit (Ultrasound Ultra Sonic, Gnatus, São Paulo, Brazil) with a stainless steel size 15 K-file, 0.02 taper (Dentsply Maillefer), at high, oscillating in the vestibule-lingual direction, inserted in the root canal 2 mm short of the WL. Firstly, 3 activations with 2.5% NaOCl were performed. Each activation lasted 20 s, in a total of 1 min per canal. Between activations, the solution was replenished. Later, the 17% EDTA was activated with ultrasound for 1 min. Finally, each canal was irrigated with 2 ml of 2.5% NaOCl to neutralize the action of the EDTA.

In Group 3, the PUI was employed the same manner as described for group 2, but citric acid was used instead of the EDTA.

After each technique, the canals were aspirated with Capillary Tips (Ultradent), and dried using size 30 paper points (Dentsply Maillefer). The crowns of the teeth were removed using a cylindrical diamond bur (KG Sorensen) in a high-speed handpiece, under constant water spray. Longitudinal grooves with 2 mm deep were cut on the bucco and lingual root surfaces without damaging the inner dentine layer around the canal using a double-sided diamond disc (KG Sorensen), and split into halves using a chisel.

Evaluation of calcium hydroxide percentage

The root half with the largest amount of remaining Ca(OH)$_2$ was chosen and viewed through a 10x stereomicroscope using a cold light source (Zeiss Ste- mi SV6; Carl Zeiss, Jena, Germany). Pictures were taken, saved in tiff format and analyzed by a single blinded examiner who had previously been calibrated for the measuring procedure. The same observer read the stereomicroscopies images twice with a 1-week interval between readings. The amount of residual Ca(OH)$_2$ on the canal walls was measured in mm$^2$ using the AxioVision Program (Zeiss, Göttingen, Germany) and recorded as a percentage of the surface area of each third of the root canal (Fig. 1).

Statistical analysis

The Kruskal-Wallis test was used to compare the percentage of remaining Ca(OH)$_2$ among groups, for each third and considering the canal as a whole. The Mann-Whitney test was performed as the post hoc multiple comparison method. The General Linear Model was used to compare the amount of remaining Ca(OH)$_2$ between thirds of the same group. The significance level was set at $\alpha = 0.05$. Statistical analysis was performed with SPSS Statistics 18 software (SPSS Inc., Chicago, IL, USA).

Results

Table 2 shows percentages of remnants of Ca(OH)$_2$ in each group, for the thirds and for the canal as a whole. Remnants of medication were
found in all experimental groups regardless of the removal technique. No significant differences were observed between experimental groups (P > 0.05). However, none of the tested groups showed similarities to the control group (P < 0.05). When the percentage of residual Ca(OH)₂ was compared between thirds into same group, no significant difference was found (P > 0.05).

Table 2 – Percentage of remaining residual Ca(OH)₂ in the canal [mean and standard deviation (SD)]

<table>
<thead>
<tr>
<th>GROUP</th>
<th>Cervical</th>
<th>Middle</th>
<th>Apical</th>
<th>Total</th>
</tr>
</thead>
<tbody>
<tr>
<td>PUI/ NaOCl + EDTA</td>
<td>2.32 ± 4.8</td>
<td>8.30 ± 19.64</td>
<td>12.35 ± 24.20</td>
<td>7.66 ± 13.79</td>
</tr>
<tr>
<td>PUI/ NaOCl + Citric Acid</td>
<td>3.32 ± 8.37</td>
<td>3.93 ± 11.88</td>
<td>9.52 ± 14.50</td>
<td>5.59 ± 8.69</td>
</tr>
<tr>
<td>Control</td>
<td>0</td>
<td>0</td>
<td>0</td>
<td>0</td>
</tr>
</tbody>
</table>

Fonte: dos autores.

Discussion

In this study, complete removal of Ca(OH)₂ from the canal walls was not attained under the tested conditions, and remnants of Ca(OH)₂ were found in all experimental groups. This result is similar to the results of previous studies, which showed considerable amounts of Ca(OH)₂ on the canal walls, regardless of the removal technique used.¹²,¹³,¹⁴,¹⁵,¹⁶,¹⁷,¹⁸,¹⁹,²⁰,²¹,²²,²³,²⁴,²⁵.

Furthermore, complementation with the PUI presented no significant improvement in the removal of Ca(OH)₂. This may be explained by the fact that all groups were previously irrigated with NaOCl and instrumented in WL with a file of oscillatory movements. Balvedi et al.²⁶ (2010), using a similar method to that of this study, also observed that there was no significant difference in the removal of Ca(OH)₂ between the action of a MAF with copious irrigation and its association with the PUI. However, these results diverge from results obtained by van der Sluis et al.²⁷ (2007), which showed that the PUI with 2% NaOCl was more effective in removing Ca(OH)₂ paste from artificial root canal grooves than either syringe delivery of NaOCl or water as irrigant. The methodological difference between the studies may explain the discrepancy in results, as van der Sluis et al.²⁷ (2007) did not use a file concurrently with manual irrigation, which differs from this research.

The PUI is based on the transmission of energy from an ultrasonic oscillating instrument to the irrigant inside the root canal.²⁷ It has been shown that the irrigation material, combined with ultrasonic vibration, is directly associated with the removal of organic and inorganic debris from the root canal walls.²¹,²⁵. Thus, considering that the effectiveness of irrigation could depend on both the mechanical flushing action and the chemical irrigation capability, an attempt was made in the present study to verify the action of the PUI with chelating solutions. EDTA and citric acid are chelating solutions capable of dissolving inorganic particles due to their property of reacting with calcium ions in dentine and forming soluble calcium chelates.²⁸ This property justifies the results of Çalt and Serper²¹ (1999) that showed complete removal of Ca(OH)₂ from the root canal after irrigation with EDTA followed by NaOCl in comparison to NaOCl alone. Rodig et al.²⁶ (2010) had already associated chelating solutions with the PUI and attained better results with citric acid and EDTA in the removal of Ca(OH)₂ in comparison to NaOCl. In this study, the best results were observed in the groups with chelating solutions and PUI (Groups 2 and 3), and the lowest average in the group using the citric acid as irrigation material (Group 3), although with no significant statistical difference (p > 0.05).

The inexistence of statistical difference between the groups in this study may be partially attributed to the high standard deviation values. The sample was composed of teeth with oval root canals, which may hinder cleaning and lead to discrepant values of remaining Ca(OH)₂ in the experimental groups, thus providing high standard deviation values. Even when a big difference is observed in the mean percentage of remaining Ca(OH)₂ between the groups, a high standard deviation may mask a possible statistical difference. Although the authors of this study have standardized the selection of specimens as much as possible, premolar teeth have shown great anatomical complexity,²⁹ which often cannot be verified through radiographic examination used during teeth selection.

Although no statistically significant difference was observed among the groups, regarding the thirds a reduction of remaining Ca(OH)₂ was observed in the cervical third, which may be explained by the action mechanism of PUI. On the other hand, the apical third had higher amounts of remaining Ca(OH)₂ even using the PUI. This may be explained due to the reduced enlargement of apical third (#30), which may have influenced the amplitude of oscillation of the instrument in the PUI.²⁰ In relation to the middle third, further reduction of Ca(OH)₂ was
observed in Group 3. This fact may be associated with anatomical difference in this third.

Different methods have been used to assess the amount of Ca(OH)\(_2\) remaining\(^{22,31,32}\). In the present study, the specimens were longitudinally split, and the percentage of residual Ca(OH)\(_2\) was calculated from images obtained under magnification of the entire area desired at stereomicroscopy. Four different aspects of the tooth were assessed: the apical, middle, and coronal thirds, as well as the entire canal. This method has been reported as effective in determining the amount of Ca(OH)\(_2\) residue and minimized subjectivity in the scoring method based on a scale\(^{16}\). The assessment of the remaining Ca(OH)\(_2\) could have been performed using scanning electron microscopy (SEM)\(^{32,38}\). However, as the aim was not to evaluate small areas and the opening of dentinal tubules, SEM was not used. Recently, micro-computed tomography has been used to assess the efficacy of different techniques of Ca(OH)\(_2\) removal. However, access to the equipment needed for this kind of assessment is limited\(^{22,34}\).

Considering that many of the samples showed remnants of Ca(OH)\(_2\), further research is necessary in order to elucidate irrigation protocols for removal of intracanal medication before final filling of the root canal to reduce side effects and optimize the benefits of Ca(OH)\(_2\) in endodontic treatment.

**Conclusion**

Within the limitations of this study, no technique was able to completely remove Ca(OH)\(_2\) from the oval root canals. The association of ‘master apical file’ with the PUI using NaOCl and chelating solutions for removing Ca(OH)\(_2\) showed similar results. Any of the techniques is suggested, since all of them were equally effective in removing Ca(OH)\(_2\).

**Resumo**

Objetivo: o objetivo deste estudo foi avaliar a efetividade da irrigação ultrassônica passiva (IUP) com diferentes soluções para remover o hidróxido de cálcio (Ca(OH)\(_2\)). Materiais e método: 48 pré-molares inferiores monorradiculares foram preparados e preenchidos com Ca(OH)\(_2\). Após 15 dias, os espécimes foram distribuídos aleatoriamente em quatro grupos, de acordo com o método de remoção da medicação intracanal (n=12). No Grupo 1, a remoção do Ca(OH)\(_2\) foi realizada, associando a irrigação convencional com NaOCl 2,5% e EDTA 17% e a lima memória no comprimento da raiz. Os Grupos 2 e 3 foram tratados da mesma maneira que o Grupo 1, seguindo da IUP com NaOCl 2,5% e EDTA 17%, e NaOCl 2,5% e ácido cítrico 10%, respectivamente. O Grupo 4 serviu como grupo controle e não recebeu qualquer medicação intracanal. As raízes foram sulcadas longitudinalmente e divididas em duas partes. A metade com a maior quantidade de remanescente foi analisada em um estereomicroscópio, e a porcentagem de área de superfície coberta com remanescente de Ca(OH)\(_2\), em relação à área de superfície total e à área de superfície por terços do canal falou foi calculada. Os resultados foram analisados estatisticamente com o teste Kruskal-Wallis e o teste Mann-Whitney. A hipótese nula foi rejeitada em 5%.

Resultados: nenhum dos grupos testados foi similar ao grupo controle (P < 0,05). Entretanto, não foi observada diferença estatística entre as técnicas de remoção do Ca(OH)\(_2\), do canal radicular. Conclusão: nem a irrigação com seringa, nem a combinação da irrigação com a seringa com IUP foram métodos efetivos para remoção do Ca(OH)\(_2\) do canal radicular.


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